



IR Spectra of Enstatite with Lattice Defects

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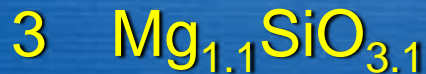
Overview of crystallization experiments



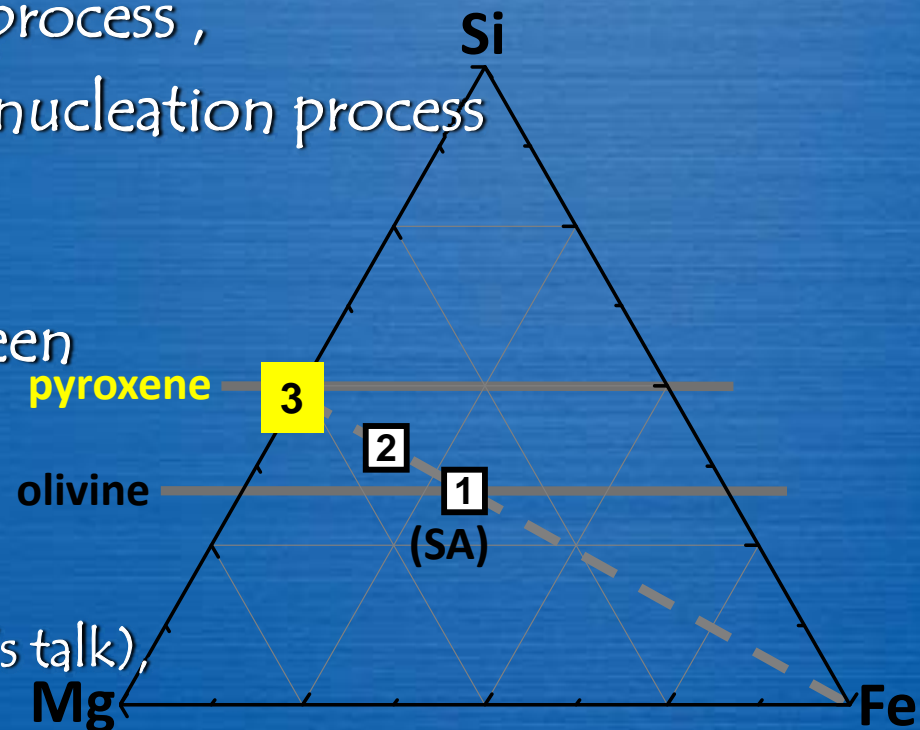
- Formulation of crystallization process ,
- Crystallization dependency on nucleation process



- Fractionation of Mg / Fe between crystal & amorphous phases



- Activation energy (→ Tsuchiyama's talk),
- Implication to observations



1 $Mg_{1.1}Fe_{0.9}SiO_4$ (Murata et al. 2007)

Formulation of Crystallization

C_{IR} : Degree of crystallization

$$\kappa_{im} = (1 - C_{IR})\kappa_{as} + C_{IR}\kappa_{xt}$$

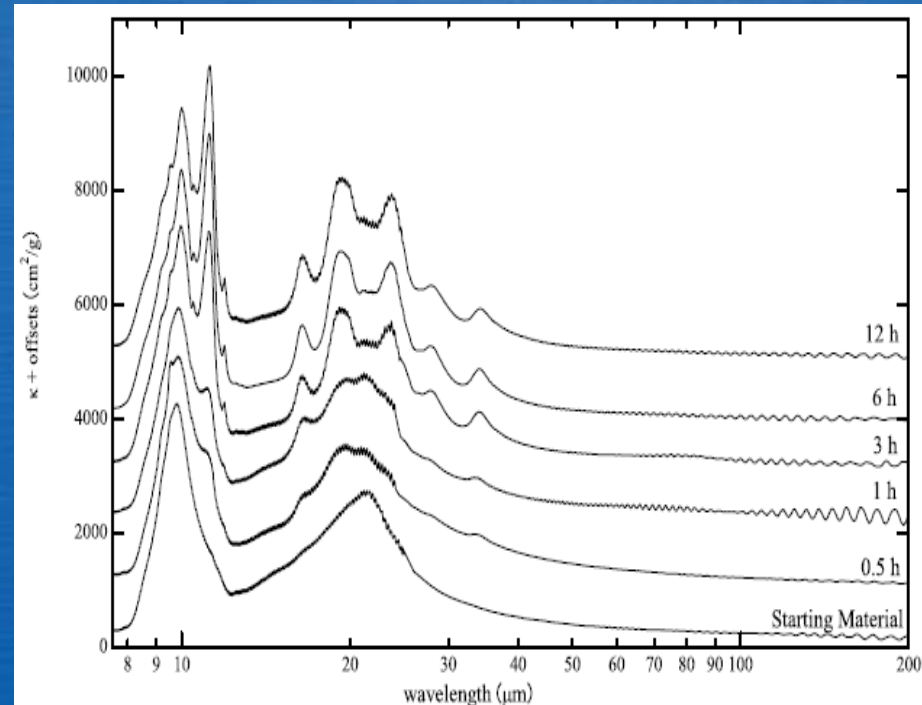


Fig. 7.—IR spectra of the samples heated at a constant temperature (680°C) for different durations.

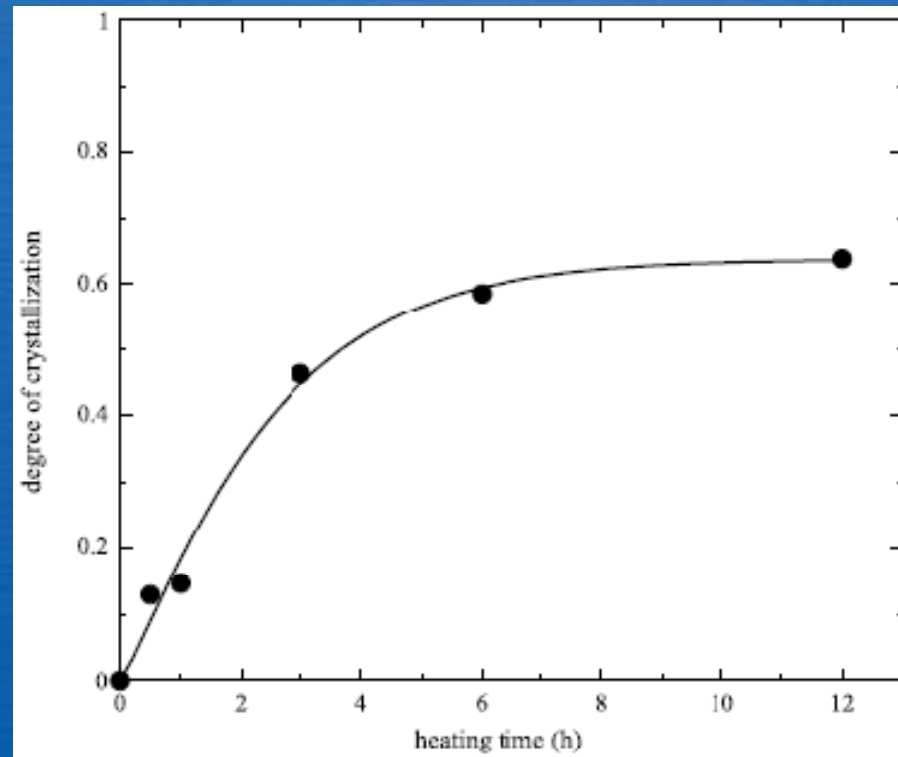
1 $Mg_{1.1}Fe_{0.9}SiO_4$ (Murata et al. 2007)

⊙ Time evolution of crystallization
JMA (Johnson-Mehl-Avrami) eq.

$$C_{IR} = C_{\infty} \{1 - \exp[-(t/\tau)^n]\}$$

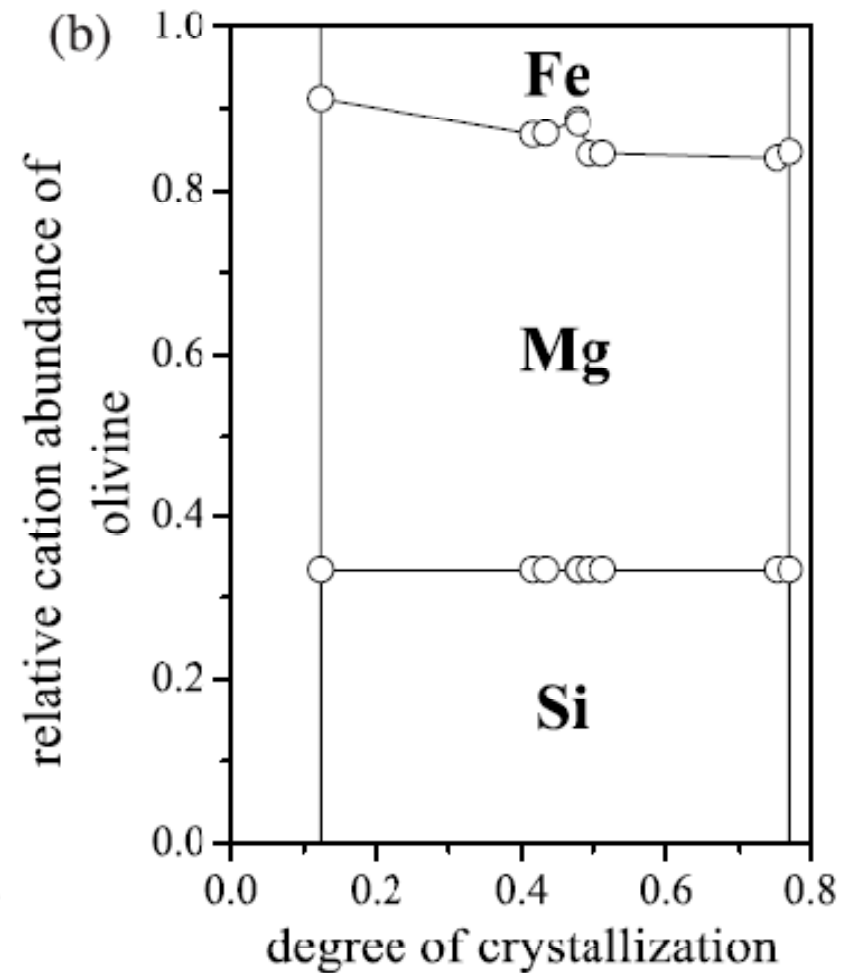
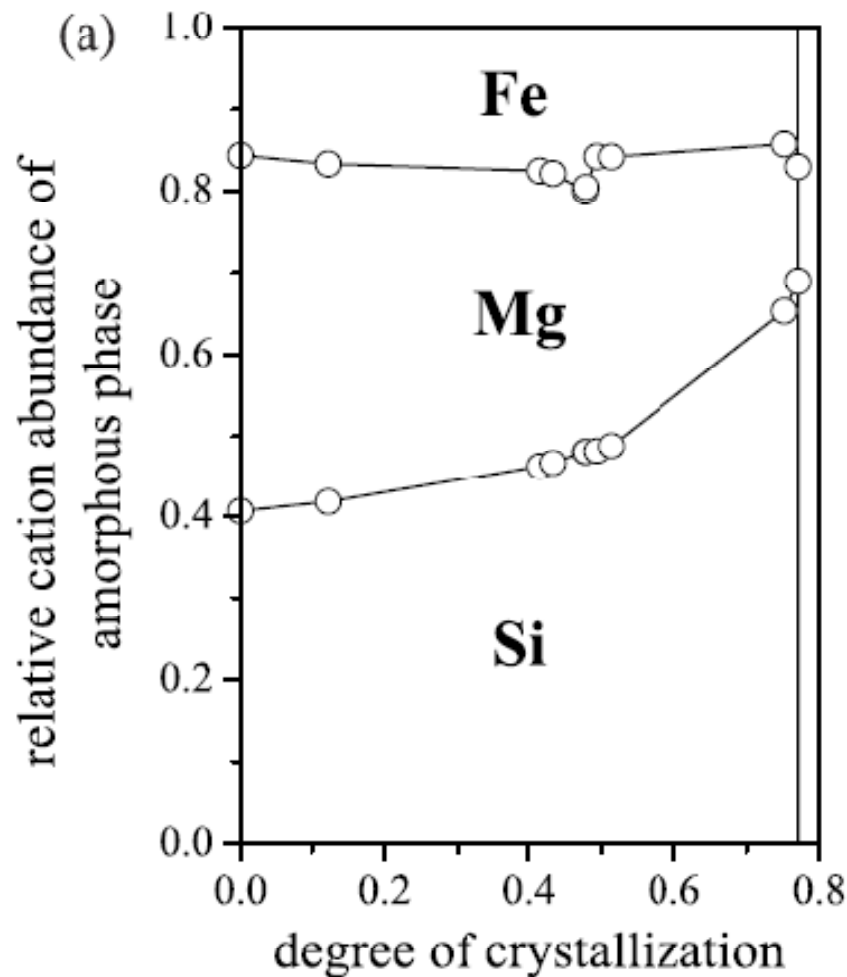
$n=1.5$: constant number of nuclei

$n=2.5$: constant nucleation rate

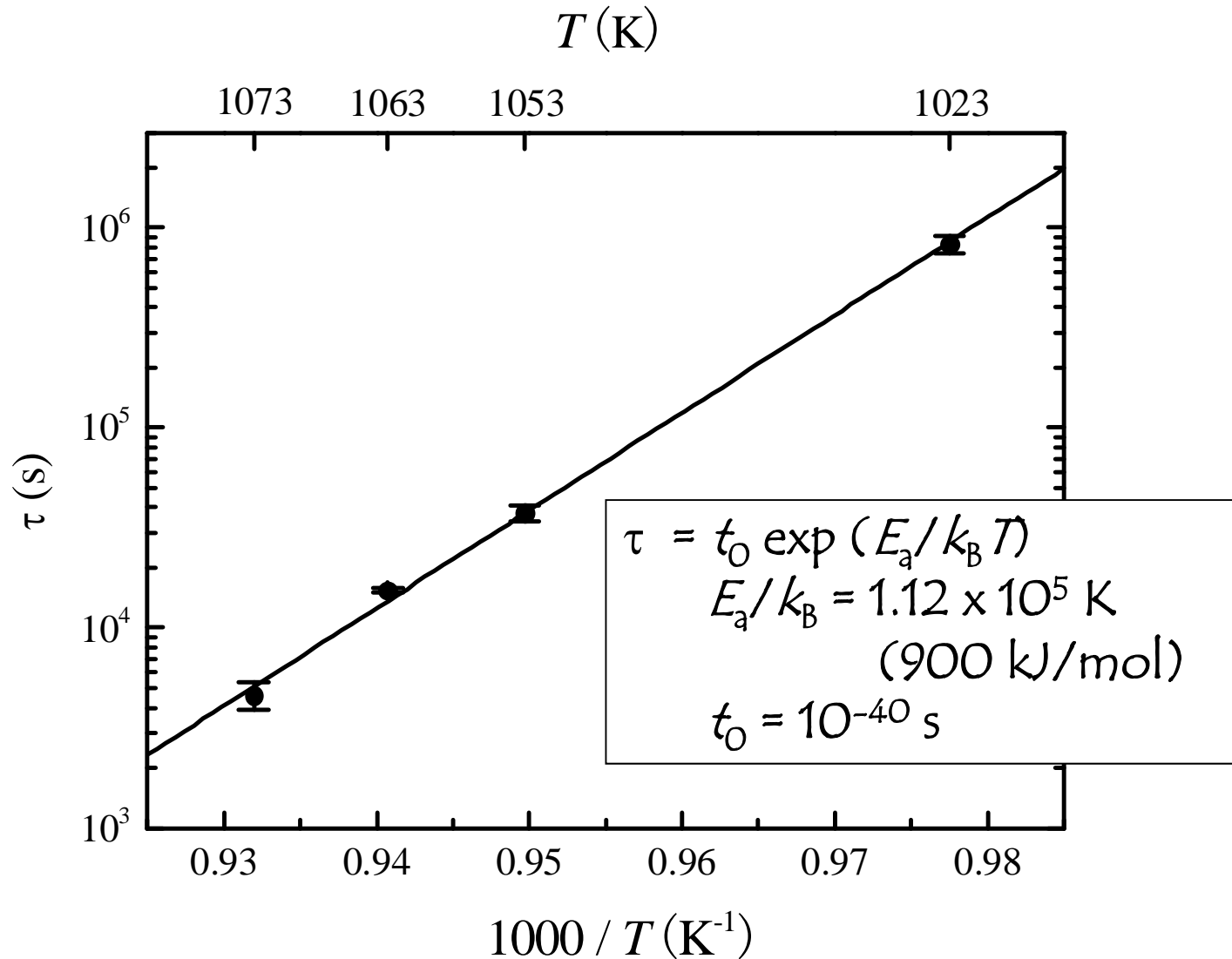


2 $Mg_{1.1}Fe_{0.4}SiO_{3.5}$ (Murata et al. 2009a)

fractionation of Mg/Fe between crystal & amorphous phases



3 $Mg_{1.1}SiO_{3.1}$ (Murata et al. 2009b)

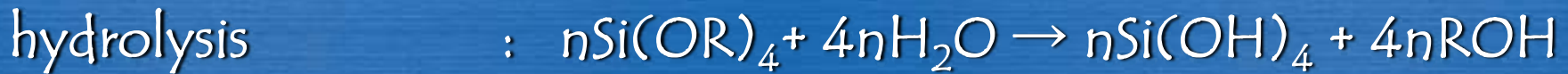




3 $Mg_{1.1}SiO_{3.1}$ (Murata et al. 2009c) Chondritic Enstatite with lattice defects

Sample preparation

Amorphous silicate ($Mg/Si = 1.07$) with Si-alkoxide (TEOS) by **Sol-Gel method**



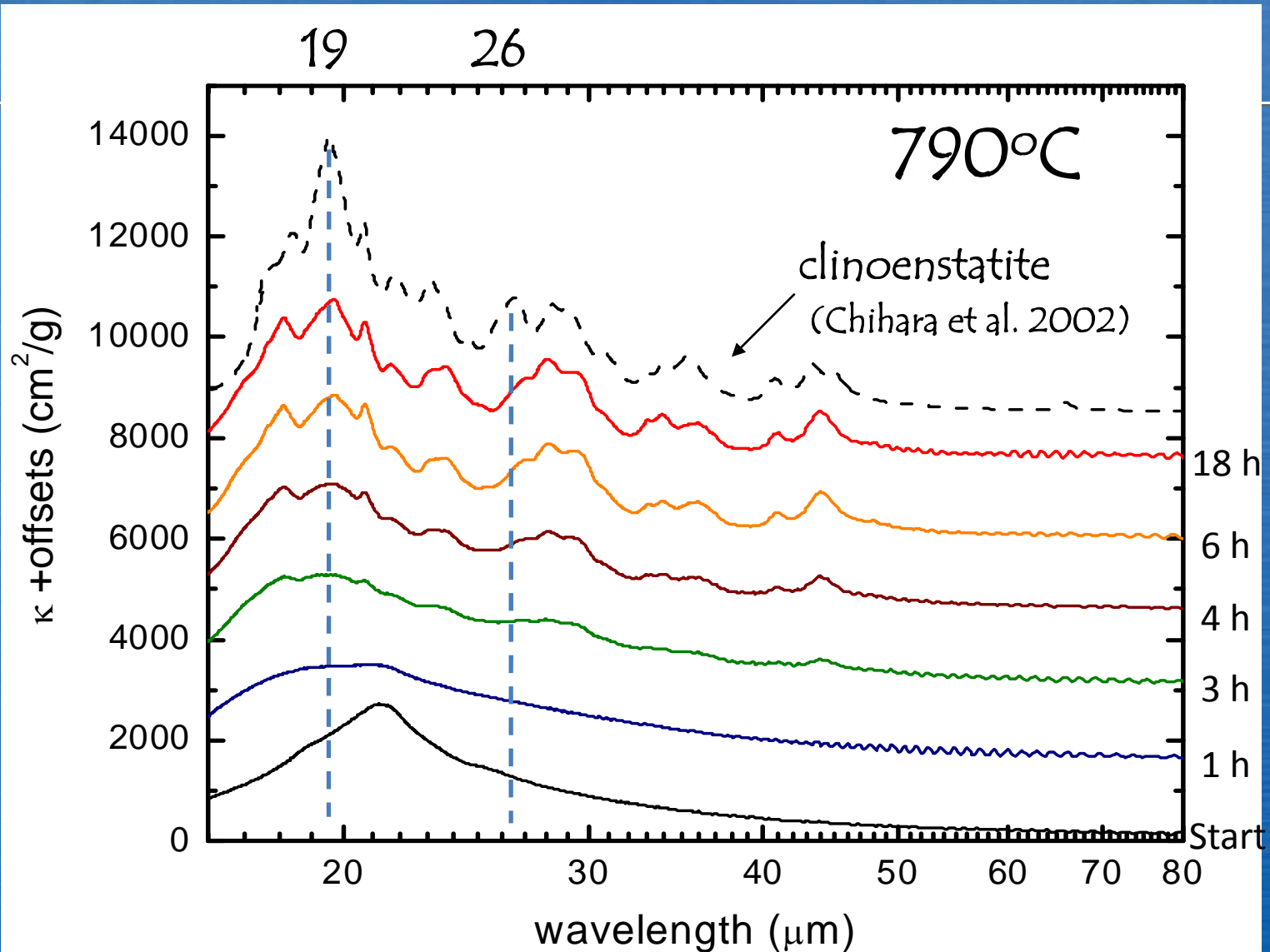
Heating at 790°C \rightarrow Crystallized Enstatite



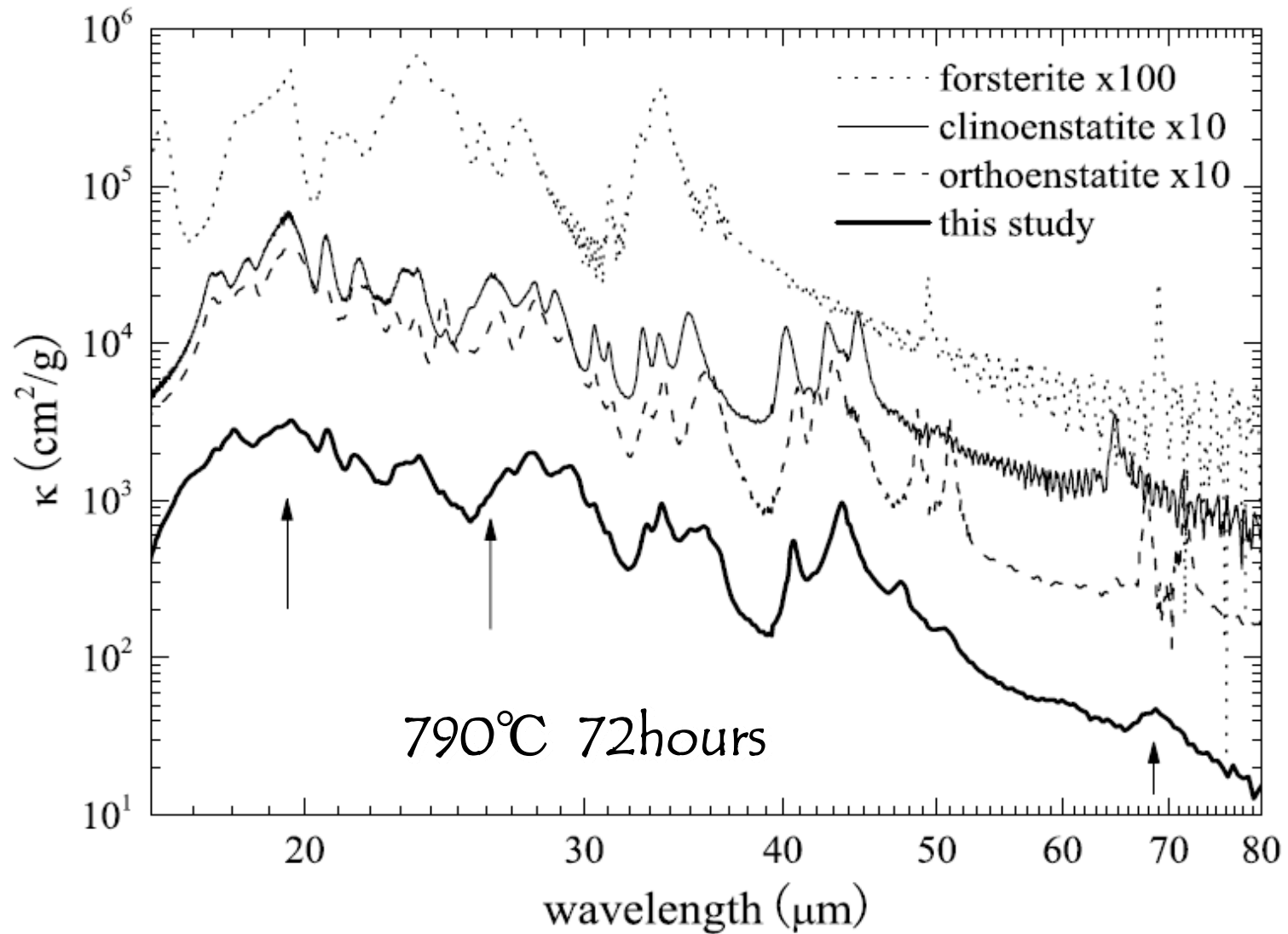
Advantages of Sol-Gel method

- ④ Very promising method to produce functional glass materials in industrial fields !
 - ☞ Homogeneity within the molecular scale
 - ☞ Required temperature is low
 - ☞ Molding is possible
 - ☞ Extremely porous → chemically active

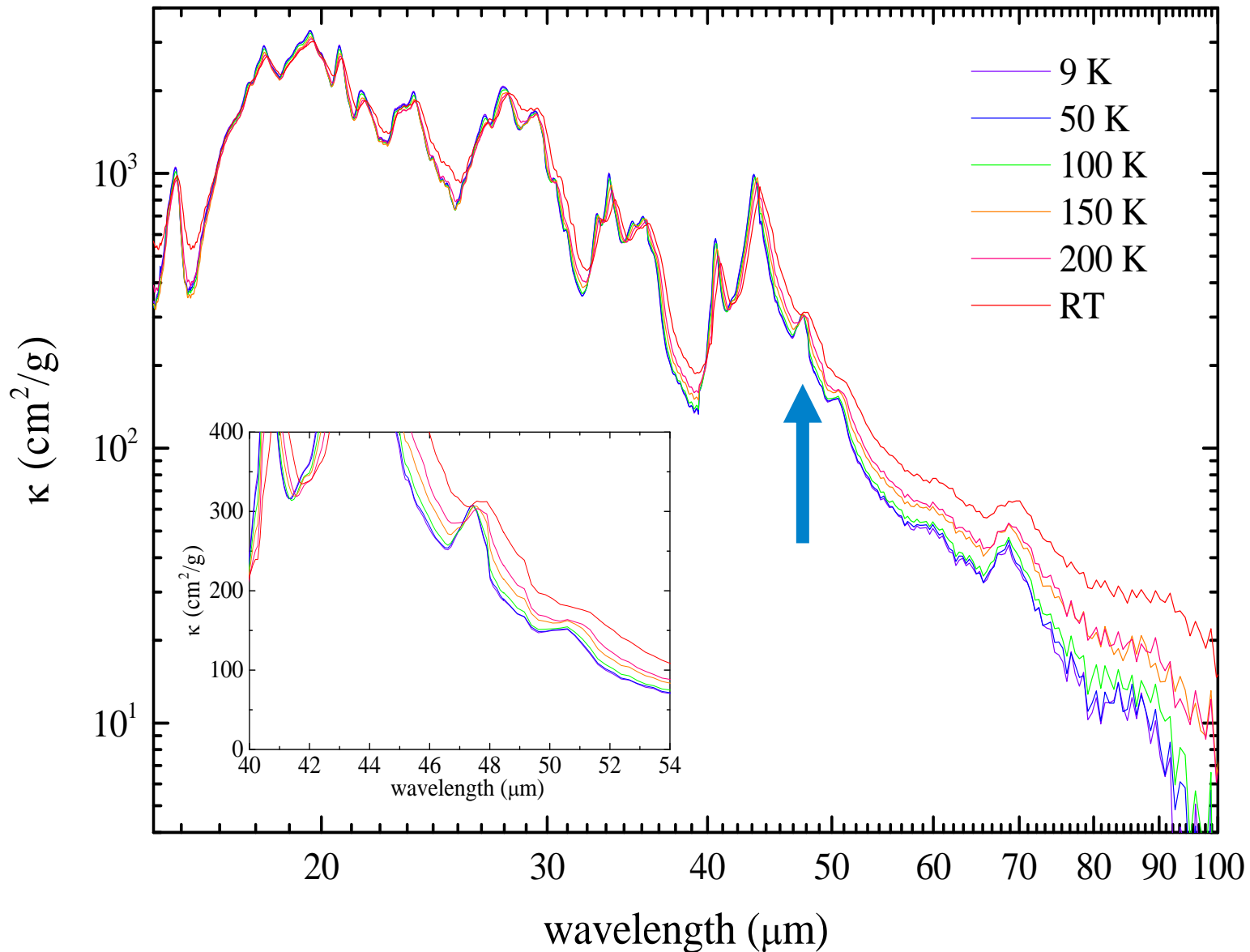
Time evolution of crystallization



Spectra at room temperature & comparison with single crystals



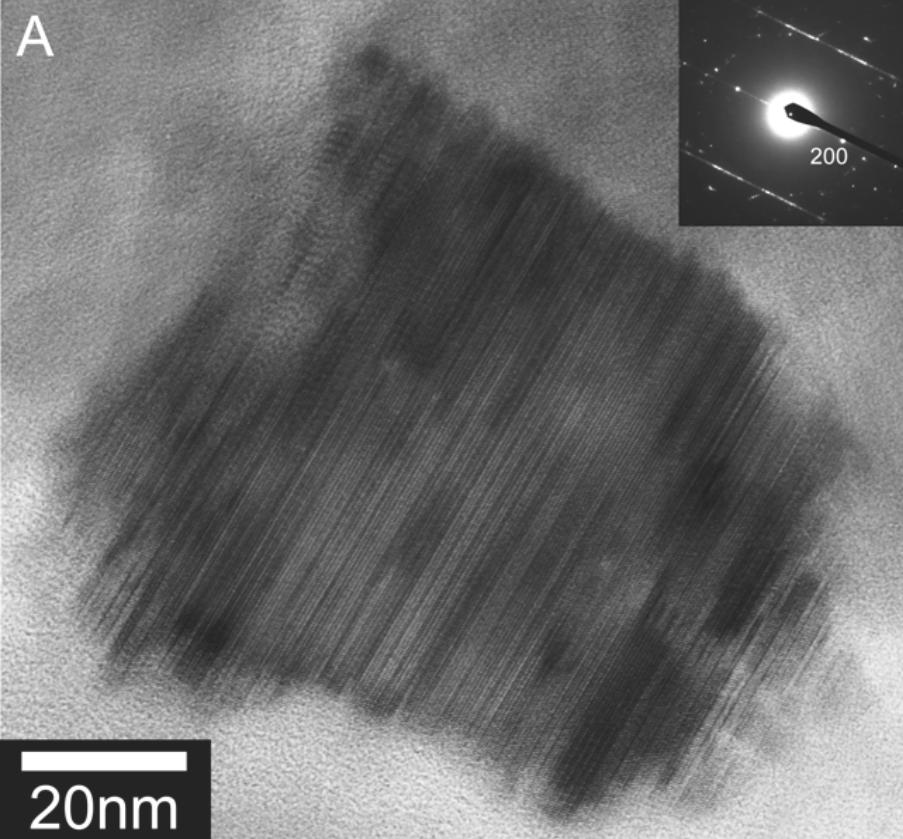
Spectra at lower temperature





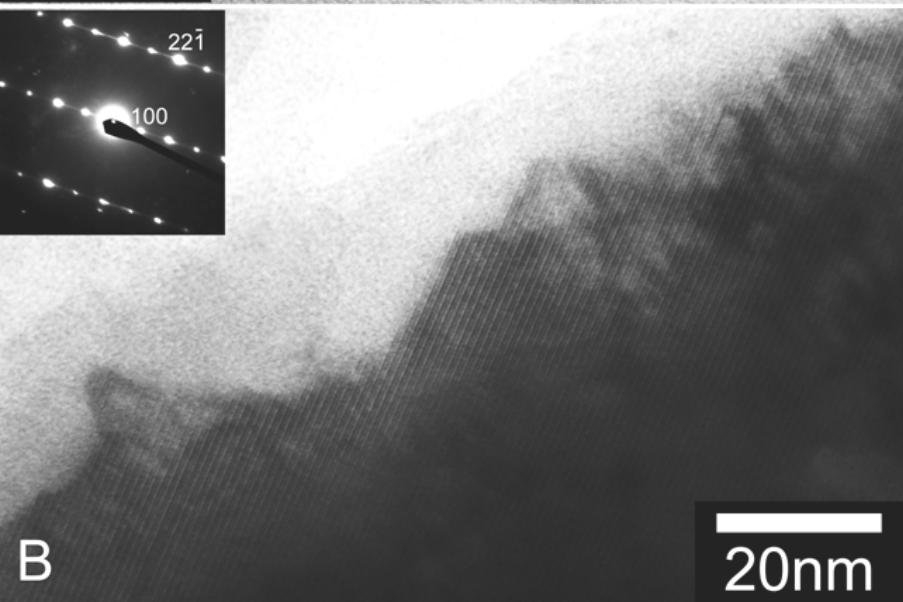
Summary of spectral features

- ④ Far-IR features are different from both of ortho- and clino- enstatite.
- ④ some peak strengths are weaker than those of single crystals (19, 26 micron)
- ④ at lower temperature prominent feature appear at 48 micron
- ④ Broad 69um feature



A crystallized enstatite
from sol-gel material

stacking faults along to
(100) direction



B clinoenstatite
single crystal

Implication to observations

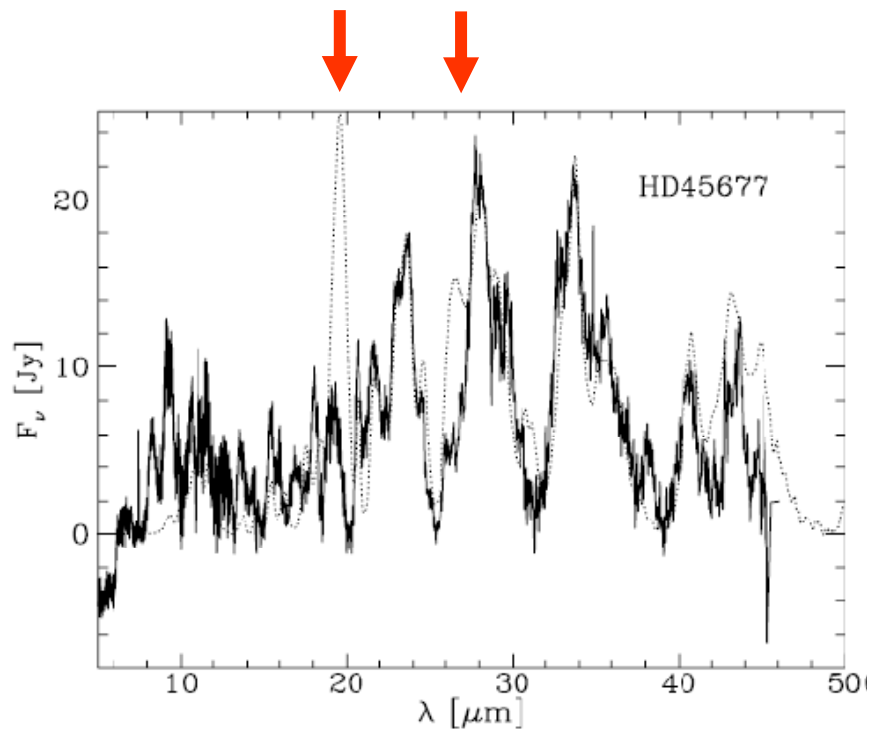


Fig. 10. A fit (dotted line) to the continuum and amorphous silicate subtracted spectrum (solid line) of HD 45677. $T_{\text{f}} = 140$ K and $T_{\text{e}} = 140$ K.

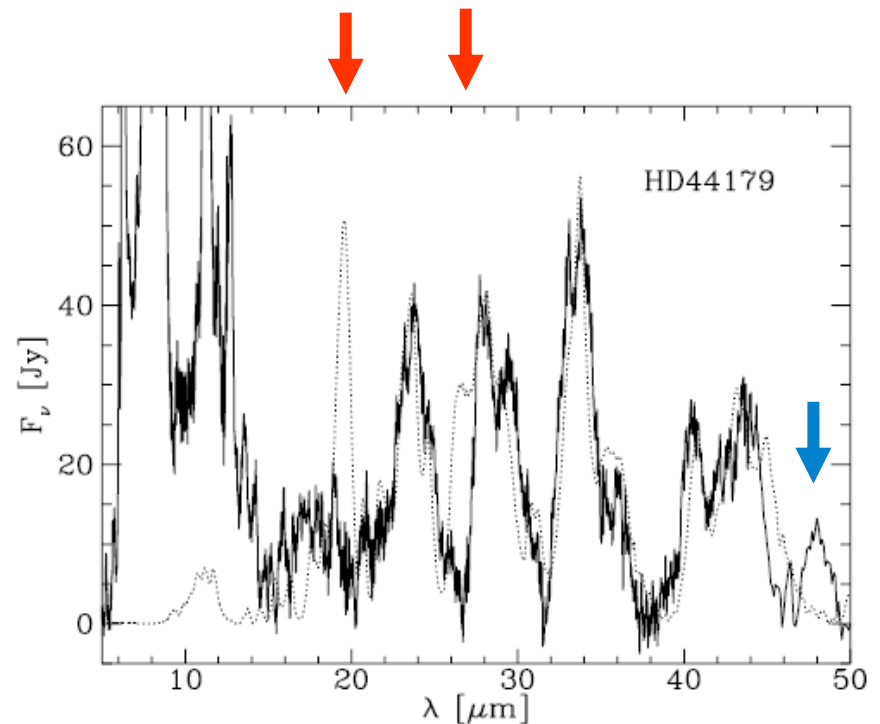


Fig. 13. A fit (dotted line) to the continuum subtracted spectrum (solid line) of HD 44179. $T_{\text{f}} = 135$ K and $T_{\text{e}} = 135$ K. Below $15 \mu\text{m}$ the spectrum is dominated by PAH features.

Previous suggestion on 48 μm unidentified feature

Ferrarotti et al. (2000) proposed that FeSi may be possible carrier of the unidentified the 48 μm feature under a peculiar abundance condition.

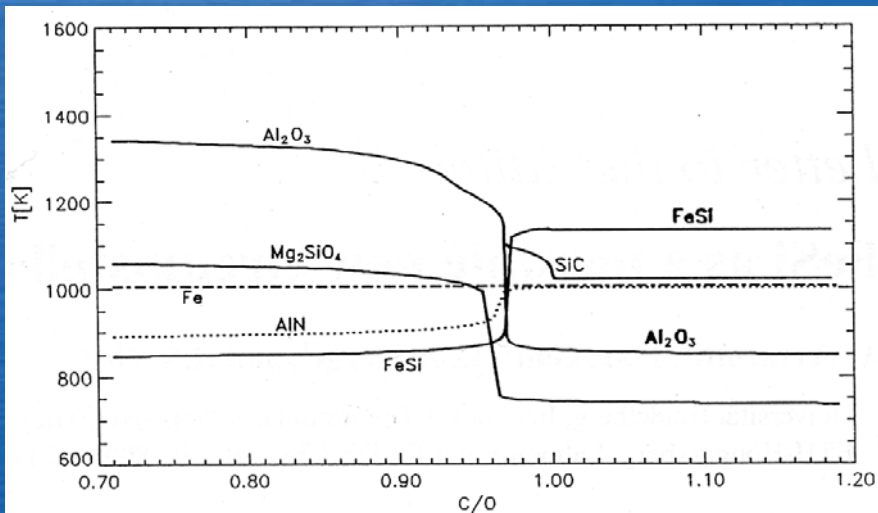


Fig. 1. Stability limits of condensates of some of the most abundant elements for varying C/O abundance ratio. The pressure is fixed at $P = 10^{-4} \text{ dyn cm}^{-2}$.

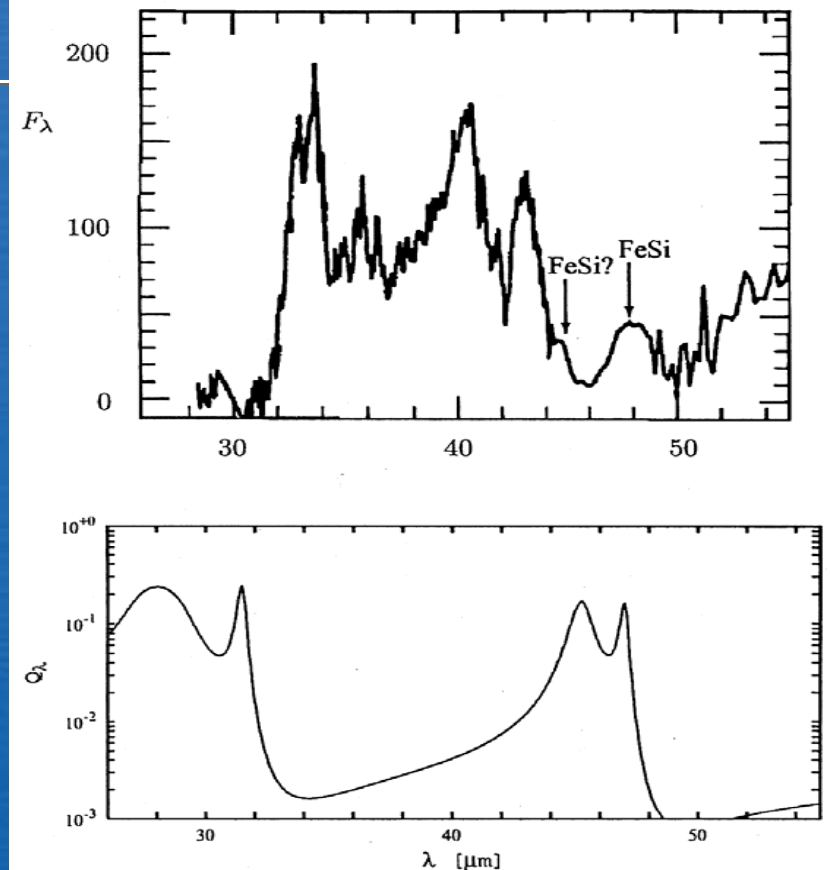


Fig. 4. *Upper Part:* Spectrum of AFGL 4106 (taken from Molster et al. 1999) in the far infrared wavelength region $28 \lesssim \lambda \lesssim 54 \mu\text{m}$. The strong bands are due to forsterite and enstatite. *Lower part:* Absorption efficiency Q_λ of FeSi grains with $0.1 \mu\text{m}$ radius at $T_{\text{dust}} = 120 \text{ K}$. A broad distribution of grain radii would broaden the absorption bands.

Comparison with previous templates

HD44179

MWC922

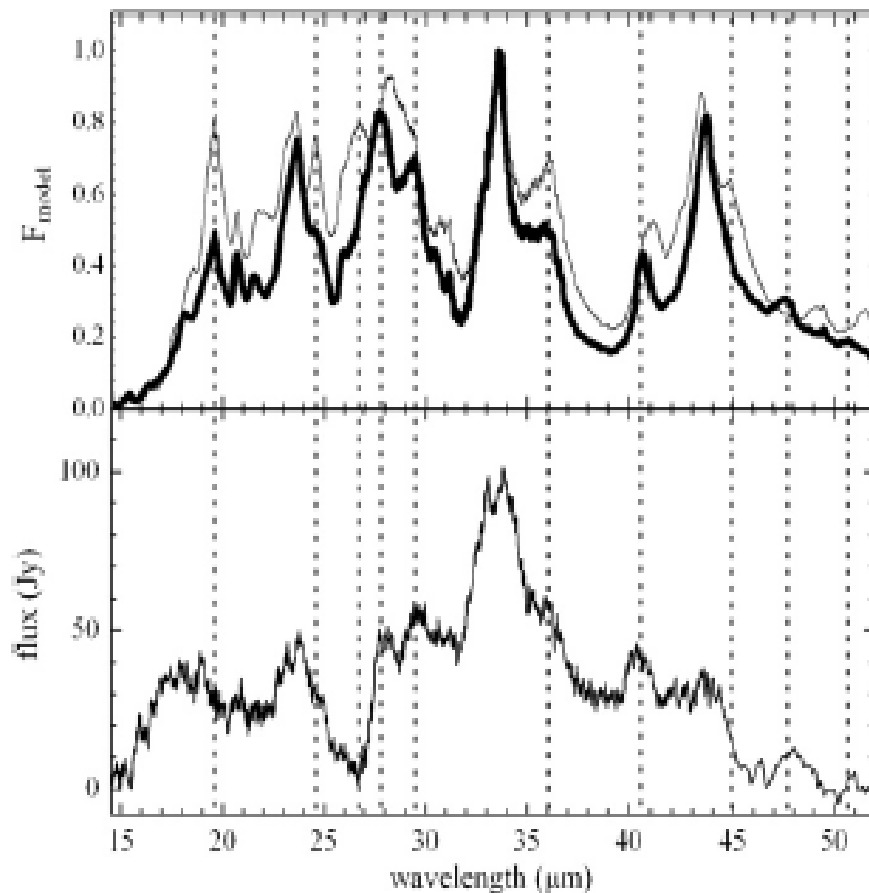


Figure 4. Comparison of the continuum-subtracted spectrum of HD 44179 with the model (1) (SCE + forsterite, thin line) and (2) (HAS + forsterite, thick line) multiplied by the Planck function of 150 K and normalized to 1.

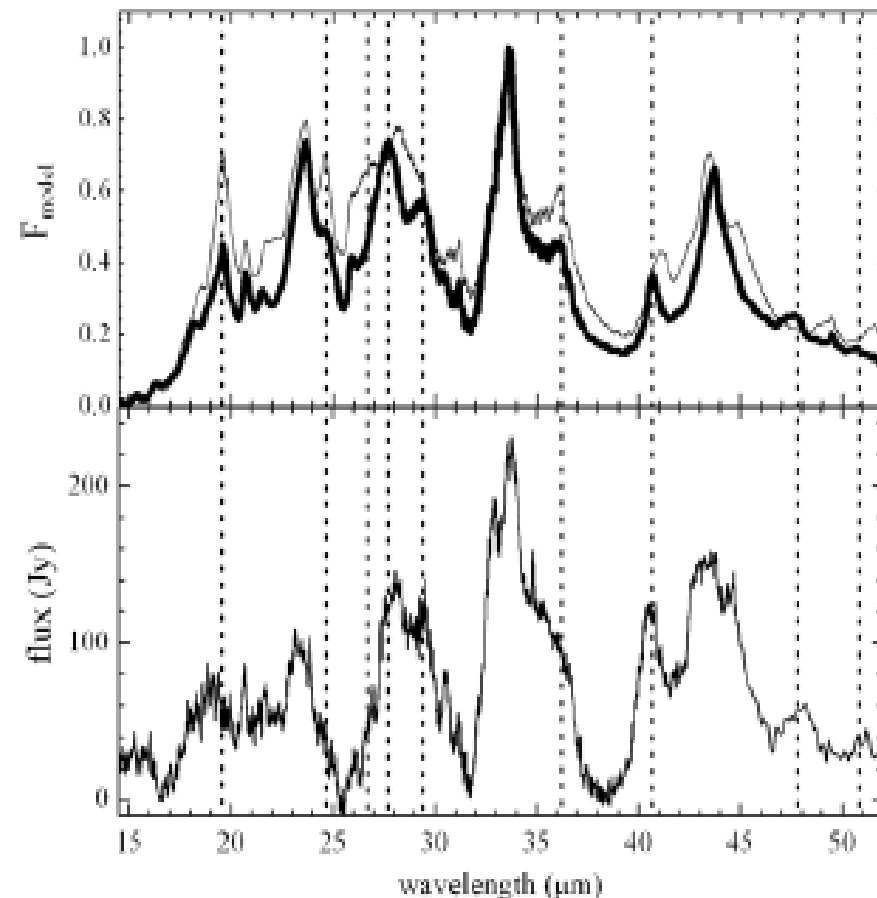
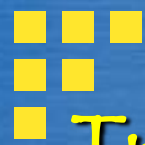
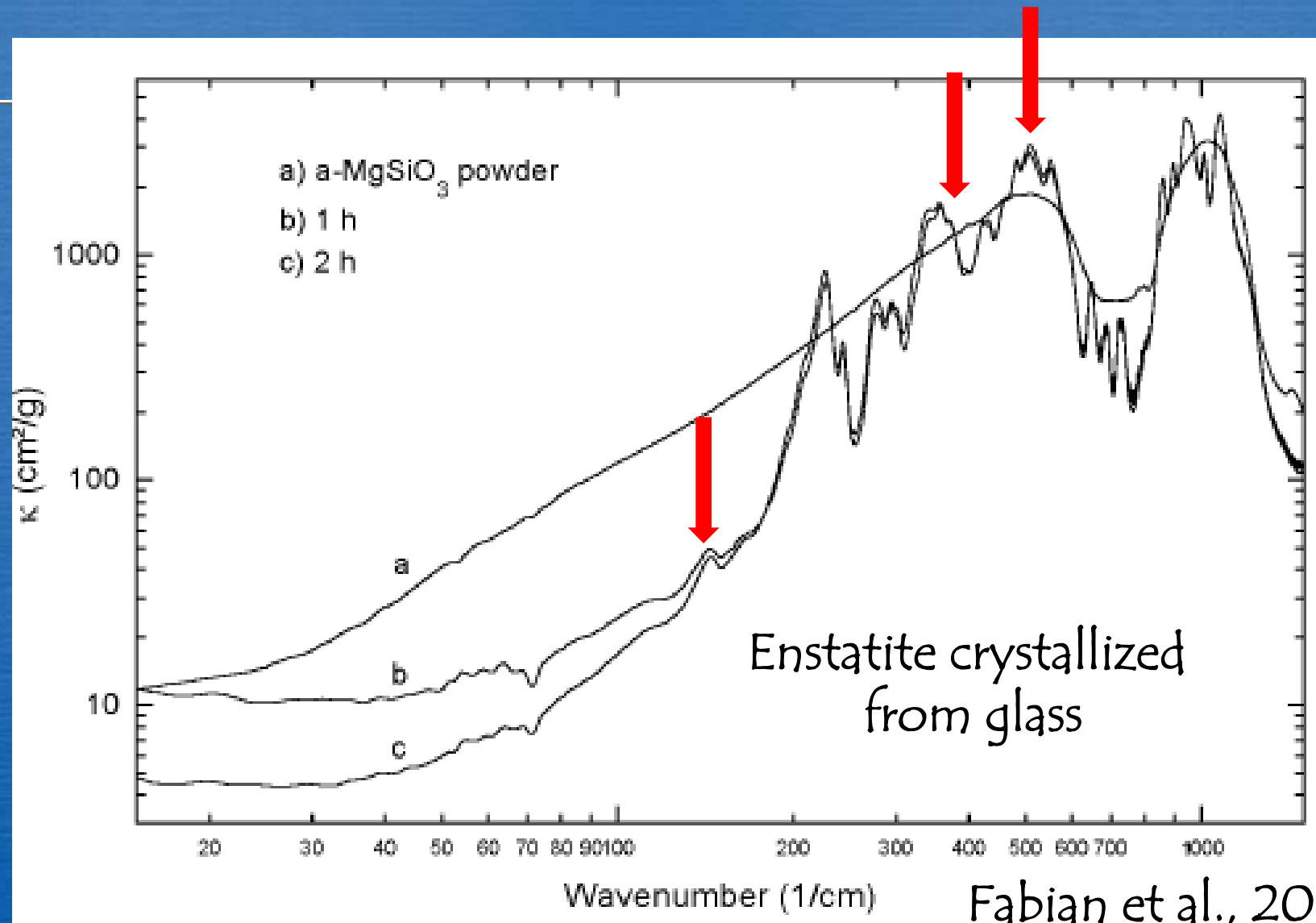
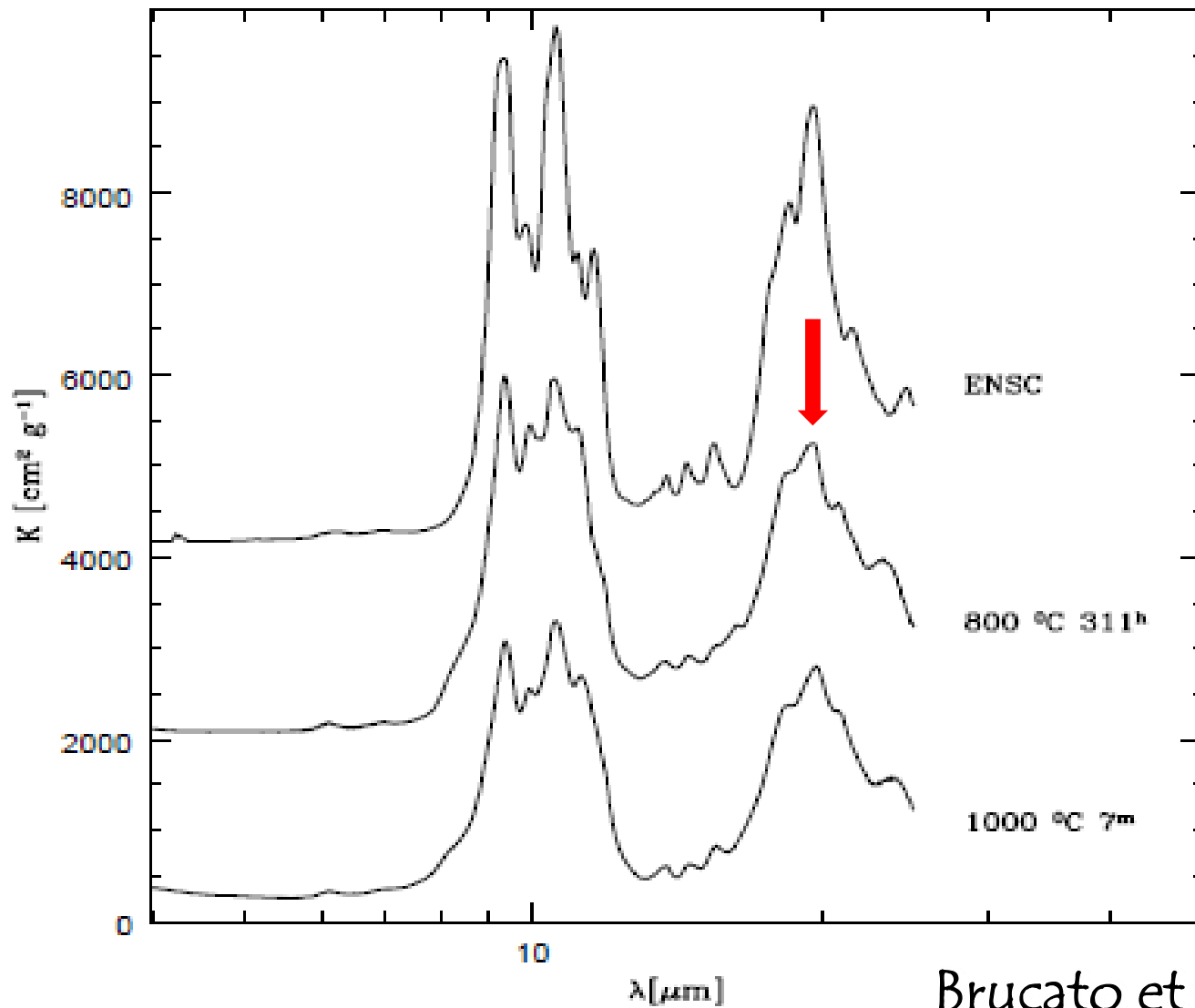


Figure 5. Comparison of the continuum-subtracted spectrum of MWC 922 with the model (1) (SCE + forsterite, thin line) and (2) (HAS + forsterite, thick line) multiplied by the Planck function of 150 K and normalized to 1.



Traces of similar features in previous studies






Brucato et al., 1999



Conclusion

- ④ IR spectrum of the sample shows different features from those of single crystals (in particular at 19, 26, 48 & 69 μm)
 - ④ Lattice defects (stacking faults) is thought to be the cause of the spectral discrepancies
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- ④ Existence of enstatite with lattice defects is suggested in circumstellar environment.